

(1.00) of each spectrum given in parentheses: MeOH 254 sh, 290 (1.00), 340 (1.00); + NaOMe: 294 sh, 326 (1.00); + AlCl₃: 261 (0.63), 303 (0.83), 324 sh, 372 (1.00); + AlCl₃ + HCl: 256 (0.70), 302 (0.84), 322 sh, 363 (1.00); + NaOAc: 290 (1.00), 330 (0.94); + NaOAc + H₃BO₃: 289 (1.00), 336 (0.73). MS *m/z* (rel. int.): 374 [M]⁺ (90) (found: 374.1002; calc. for C₁₉H₁₈O₇: 374.1005), 373 [M - 1]⁺ (14), 359 [M - Me]⁺ (100), 197 [A - 15]⁺ (38) (found: 197.0086; calc. for C₈H₅O₆: 197.0085). ¹H NMR (350 MHz, DMSO-*d*₆): δ 7.69 (1H, *dd*, *J* = 8.4 and 2 Hz, H-6'), 7.58 (1H, *d*, *J* = 2 Hz, H-2'), 7.18 (1H, *d*, *J* = 8 Hz, H-5'), 6.98 (1H, *s*, H-3), 3.97 (3H, *s*, OMe), 3.94 (3H, *s*, OMe), 3.89 (3H, *s*, OMe), 3.87 (3H, *s*, OMe). *R_f* 0.50 on Polyamide DC6 (Merck), C₆H₆-petrol (bp 100–140°)–MeOH–MeCOEt (5:5:1:1) and 0.22 on cellulose in 30% HOAc.

Acknowledgements—We thank M. Waton (CNRS, Solaise, France) for NMR spectra and Drs. Rodriguez (Madrid), Tatum (Winter Haven) and Tomas (Murcia) for gifts of flavonoid samples. This research, partially supported by Ministère de l'Industrie et de la Recherche, France, Comité D.M.D.R. No. 8360846, is part of the programme Menthés realized in collaboration with Dr. R. Perrin (Laboratoire de Chimie Industrielle, Univ. Lyon), Dr. A. Perrin (Laboratoire de Biologie Végétale, Univ. St Etienne), Dr. Poupet (I.N.R.A., Antibes) and M. Lamy (Chambre d'Agriculture de la Drôme).

REFERENCES

- González, A. G., Fraga, B. M., Hernandez, M. G., Larruga, F., Luis, J. G. and Ravelo, A. G. (1978) *Lloydia* **41**, 279.
- Adzet, T. and Martinez, F. (1981) *Biochem. Syst. Ecol.* **9**, 293.
- Rodriguez, B. (1977) *Phytochemistry* **16**, 800.
- Tomas, F., Ferreres, F. and Guirado, A. (1979) *Phytochemistry* **18**, 185.
- Van Den Broucke, C. O., Domisse, R. A., Esmans, E. L. and Lemli, J. A. (1982) *Phytochemistry* **21**, 2581.
- Voirin, B. (1983) *Phytochemistry* **22**, 2107.
- Miski, M., Ulubelen, A. and Mabry, T. J. (1983) *Phytochemistry* **22**, 2093.
- Van Den Broucke, C. O., Lemli, J. and Lamy, J. (1982) *Plant. Med. Phytoth.* **XVI**, 310.
- Gella, E. V. (1966) *Fenofnye Soedin Ikh. Biol. Funkts*, Mater Vses Simp. 1st.
- Gella, E. V., Makarova, G. V. and Borisyuk, Y. G. (1967) *Farm. Zh. (Kiev)* **22**, 80.
- Lallement-Guilbert, N. and Bezanger-Beauquesne, L. (1970) *Plant. Med. Phytoth.* **4**, 92.
- Kohlmunzer, S., Gryzbek, J. and Sodzawiczny, K. (1975) *Herb. Pol.* **21**, 130.
- Palutova, T. P. (1973) *Bio. Zh.* **17**, 17.
- Zakharova, O. I., Zakharov, A. M., Glyzin, V. I. and Smirnova, L. P. (1982) *Khim. Prir. Soedin.* **5**, 652.
- Zakharova, O. I., Zakharov, A. M., Smirnova, L. P. and Kovineva, V. M. (1983) *Khim. Prir. Soedin.* **5**, 645.

Phytochemistry, Vol. 23, No. 12, pp. 2973–2975, 1984.
Printed in Great Britain.

0031-9422/84 \$3.00 + 0.00
© 1984 Pergamon Press Ltd.

STRUCTURAL REVISION OF THE FLAVONE MAJORANIN FROM *MAJORANA HORTENSIS*

BERNARD VOIRIN, JEAN FAVRE-BONVIN, V. INDRA* and A. G. RAMACHANDRAN NAIR*

Laboratoire de Phytochimie, U.E.R. des Sciences de la Nature, 43 Bd du 11 Novembre 1918, 69623 Villeurbanne, France; *Department of Chemistry, Jawaharlal Institute, Pondicherry 605006, India

(Received 23 May 1984)

Key Word Index—*Majorana hortensis*; Labiatae; majoranin; thymonin; 5,6,4'-trihydroxy-7,8,3'-trimethoxyflavone.

Abstract—Majoranin isolated from *Majorana hortensis* and characterized earlier as 5,7,4'-trihydroxy-6,8,3'-trimethoxyflavone was found to be different from sudachitin of the same structure. Its true structure has been established as 5,6,4'-trihydroxy-7,8,3'-trimethoxyflavone (thymonin) by spectral data and direct comparison.

INTRODUCTION

A trihydroxytrimethoxyflavone, majoranin, was isolated from the leaves of *Majorana hortensis* and assigned the structure 5,7,4'-trihydroxy-6,8,3'-trimethoxyflavone by Subramanian *et al.* [1]. Owing to the earlier occurrence of a flavone, sudachitin, with the same structure [2], Wollenweber [3] cited majoranin as a synonym. However,

there exist subtle differences in the physical constants and UV spectra of the two compounds. This necessitated a re-examination of the structure of majoranin. Thus, since the precise differentiation of the positions of hydroxyl and methoxyl groups in flavonoid compounds can be accomplished by UV measurements [4], a re-examination of the structure of majoranin was undertaken and the results

Table 1. UV spectral properties (λ_{max} , nm) of the flavones

	Sudachitin			Majoranin			Thymonin		
MeOH	256 sh	281 (0.80)	344 (1.00)	254 sh	290 (0.86)	343 (1.00)	256 sh	290 (0.83)	342 (1.00)
AlCl ₃	262 (0.61)	290 (0.73)	303 i (1.00)	412 sh (0.41)	262 (0.66)	322 i (1.00)	244 sh (0.72)	302 (0.91)	380 (1.00)
AlCl ₃ + HCl	260 (0.64)	292 (0.77)	302 i (1.00)	412 sh (0.41)	258 (0.77)	320 i (0.72)	244 sh (0.87)	302 (0.94)	372 (1.00)
NaOMe	260 sh	284 (0.93)	340 sh (1.00)	412 (1.00)	266 (0.46)	324 i (1.00)	270 (0.80)	308 (0.79)	407 (1.00)
NaOAc		282 (1.00)	322 (0.52)	404 (0.73)	266 (0.91)	330 sh (1.00)	260 (0.93)	284 sh (0.93)	400 (1.00)

The relative absorbance is presented for the λ_{max} of each compound, using the highest peak as 100% (1.00) of each spectrum. sh, Shoulder; i, inflection.

led to the establishment of its structure as 5,6,4'-trihydroxy-7,8,3'-trimethoxyflavone.

RESULTS AND DISCUSSION

The compound is, in agreement with the data presented by earlier workers [1], a trihydroxytrimethoxyflavone with a free 5-hydroxyl group in the fully oxygenated A-ring and a B-ring with 4'-hydroxyl and 3'-methoxyl groups. The UV spectral data (Table 1; MeOH: λ_{\max} , band II; and MeOH + AlCl₃ + HCl: band I, one peak, 350–381 nm) clearly indicate that the remaining hydroxyl group is located at C-6 [4] and not at C-7 as previously assigned [1]. The absence of any shift of band II in the NaOAc spectrum also supported this finding, though caution has to be exercised in drawing this conclusion in the case of flavonoids substituted at C-6 and/or C-8 [5]. The absence of band III in the NaOH spectrum further supports the absence of a 7-hydroxyl group (presence of 7-methoxyl) though there are exceptions [6]. The mass spectrum of the compound exhibited the parent ion at m/z 360 (78%) with the base peak at m/z 345, in agreement with the mass spectral pattern suggested for 5-hydroxy-7,8-dimethoxyflavones [7]. Thus, majoranin has been established as 5,6,4'-trihydroxy-7,8,3'-trimethoxyflavone or thymonin, a flavone recently isolated from *Thymus vulgaris* [8]. Comparison of the spectral data (UV, ¹H NMR and MS) of majoranin with those of thymonin [8] revealed the two flavones to be identical. Finally, direct comparison including co-chromatography with authentic thymonin and sudachitin confirmed the identity of majoranin with thymonin and its non-identity with sudachitin.

The name 'majoranin' appears to have priority over 'thymonin' as the flavone was first isolated from *Majorana hortensis* [1] in a crystalline form (mp 225–226°; triacetyl derivative, mp 186–187°) even though the credit for correct characterization goes to Van den Broucke *et al.* [8]. It is interesting that majoranin (thymonin) has been obtained in both cases from the Labiatae, which are known for synthesizing unusual flavonoids [9, 10].

EXPERIMENTAL

For isolation and purification of majoranin, see ref. [1]. UV spectral data for sudachitin, majoranin and thymonin are given in

Table 1. ¹H NMR spectra were recorded at 350 MHz in DMSO-*d*₆; chemical shifts in δ values with respect to internal TMS. Majoranin: δ 3.89, 3.93, 3.96 (3H each, s, OMe), 6.90 (1H, s, 3-H), 6.96 (1H, *d*, *J* = 9.0 Hz, 5'-H), 7.57 (2H, unresolved *m*, irradiation at 6.96 produces *s* at 7.50, 2'- and 6'-H). Thymonin [8]: δ 3.88, 3.92, 3.94 (3H each, s, OMe), 6.92 (2H, *m*, 3-H and 5'-H), 7.52 (2H, *m*, 2'-H and 6'-H).

EIMS were recorded at 70 eV; intensities are indicated as % within parentheses. Majoranin: m/z 360 [M]⁺ (78) (high resolution: 360.0843; calc. for C₁₈H₁₆O₈: 360.0845), 359 (5), 345 (100), 197 (27), 169 (10), 151 (7), 148 (5). Thymonin [8]: m/z 360 [M]⁺ (68) (high resolution: 360.0815), 359 (5), 345 (100), 197 (20), 169 (6), 151 (5), 148 (2).

PC was done on Whatman No. 1 using 15% HOAc (A) and *t*-BAW (3:1:1) (B) solvents; TLC on cellulose using 30% HOAc (C) and on polyamide using C₆H₆-petrol (bp 100–140°)-MeCOEt-MeOH (5:5:1:1) (D). *R_f*s in the order majoranin, thymonin and sudachitin were: 10, 10, 6 (A); 91, 91, 93 (B); 24, 24, 29 (C) and 23, 23, 17 (D).

Acknowledgements—We thank Dr. Van den Broucke (Belgium) and Dr. Strelisky (Budapest, Hungary) for the supply of thymonin and sudachitin, respectively; Mr. Waton (CNRS, Solaise, France) for NMR spectra, and the Director, Jawaharlal Institute, for encouragement.

REFERENCES

1. Subramanian, S. S., Nair, A. G. R., Rodriguez, E. and Mabry, T. J. (1972) *Curr. Sci.* **41**, 1547.
2. Horie, T., Masumara, M. and Okumara, F. S. (1961) *Bull. Chem. Soc. Jpn.* **34**, 185.
3. Wollenweber, E. (1982) in *The Flavonoids, Advances in Research* (Harborne, J. B. and Mabry, T. J., eds.), p. 199. Chapman & Hall, London.
4. Voirin, B. (1983) *Phytochemistry* **22**, 2107.
5. Rabesa, Z. and Voirin, B. (1978) *Phytochemistry* **17**, 180.
6. Bacon, J. and Mabry, T. J. (1976) *Rev. Latinoam. Quim.* **7**, 83.
7. Goudard, M., Favre-Bonvin, J., Strelisky, J., Nogradi, M. and Chopin, J. (1979) *Phytochemistry* **18**, 186.
8. Van den Broucke, C. O., Domisse, R. A., Esmans, L. E. and Lemli, J. A. (1982) *Phytochemistry* **21**, 2581.
9. Harborne, J. B. (1967) *Comparative Biochemistry of the Flavonoids*. Academic Press, London.
10. Harborne, J. B. and Williams, C. A. (1971) *Phytochemistry* **10**, 367.